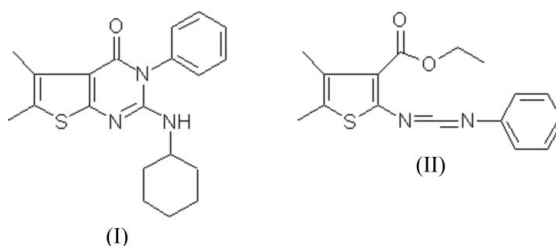
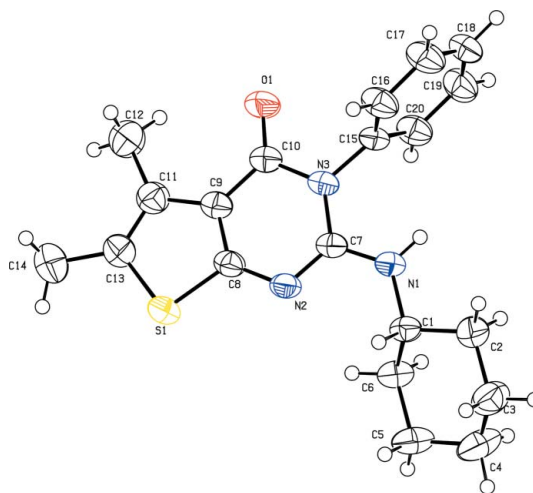


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Key indicators

Single-crystal X-ray study
 $T = 292\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.003\text{ \AA}$
 R factor = 0.048
 wR factor = 0.138
Data-to-parameter ratio = 16.6For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.2-Cyclohexylamino-5,6-dimethyl-3-phenyl-
3H-thieno[2,3-d]pyrimidin-4(3H)-oneMolecules of the title compound, $\text{C}_{20}\text{H}_{23}\text{N}_3\text{OS}$, form a
supramolecular structure *via* intermolecular $\text{C}-\text{H}\cdots\pi$
interactions.Received 3 July 2006
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Comment

Derivatives of thienopyrimidine are of great importance
because of their biological properties (Ding *et al.*, 2004). We
have recently focused our attention on the synthesis of
heterocyclic systems containing a fused pyrimidinone ring
using the aza-Wittig reaction (Hu *et al.*, 2005). The title
compound, (I), may be used as a new precursor to obtain
bioactive molecules. Its structure is reported here (Fig. 1).The bond lengths and angles are unexceptional. The two
fused rings are essentially coplanar (Table 1), with maximum
deviations of 0.056 (2) and -0.042 (2) \AA for C7 and S1,
respectively. The dihedral angle between the C15–C20 phenyl
ring and the thienopyrimidinone system is 89.00 (1) $^\circ$. The
cyclohexyl ring adopts a distorted chair conformation [$\varphi =$
356.64 (2) $^\circ$ and $\theta = 2.53$ (2) $^\circ$, and puckering amplitude =
0.575 (2) \AA ; Cremer & Pople, 1975].**Figure 1**
View of the molecular structure of (I), showing the atom-labeling scheme.
Displacement ellipsoids are drawn at the 50% probability level.

The crystal structure is stabilized by intermolecular C—H $\cdots\pi$ interactions (Fig. 2 and Table 2)

Experimental

To a solution of (II) (3 mmol) in dichloromethane (15 ml) was added cyclohexylamine (3 mmol). The reaction mixture was allowed to stand for 2 h; the solvent was then removed and anhydrous ethanol (10 ml) with several drops of EtONa in EtOH was added. The mixture was stirred for 5 h at room temperature. The solution was concentrated under reduced pressure and the residue was recrystallized from ethanol to give the title compound. (I) was recrystallized from ethanol–dichloromethane (1:2 v/v) at room temperature, yielding crystals suitable for single-crystal X-ray diffraction.

Crystal data

$C_{20}H_{23}N_3OS$	$V = 933.1 (5) \text{ \AA}^3$
$M_r = 353.47$	$Z = 2$
Triclinic, $P\bar{1}$	$D_x = 1.258 \text{ Mg m}^{-3}$
$a = 8.813 (3) \text{ \AA}$	Mo $K\alpha$ radiation
$b = 10.064 (3) \text{ \AA}$	$\mu = 0.19 \text{ mm}^{-1}$
$c = 11.526 (3) \text{ \AA}$	$T = 292 (2) \text{ K}$
$\alpha = 100.522 (5)^\circ$	Block, colorless
$\beta = 99.819 (5)^\circ$	$0.20 \times 0.20 \times 0.10 \text{ mm}$
$\gamma = 107.060 (5)^\circ$	

Data collection

Bruker SMART 4K CCD area-detector diffractometer	5533 measured reflections
φ and ω scans	3794 independent reflections
Absorption correction: multi-scan (SADABS; Sheldrick, 2003)	3111 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.964$, $T_{\max} = 0.982$	$R_{\text{int}} = 0.052$
	$\theta_{\text{max}} = 26.5^\circ$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0835P)^2 + 0.0173P]$
$R[F^2 > 2\sigma(F^2)] = 0.048$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.138$	$(\Delta/\sigma)_{\text{max}} = 0.004$
$S = 1.04$	$\Delta\rho_{\text{max}} = 0.24 \text{ e \AA}^{-3}$
3794 reflections	$\Delta\rho_{\text{min}} = -0.27 \text{ e \AA}^{-3}$
228 parameters	
H-atom parameters constrained	

Table 1

Selected torsion angles ($^\circ$).

S1—C8—N2—C7	179.00 (11)	N2—C8—S1—C13	−177.06 (14)
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Table 2

Hydrogen-bond geometry (\AA , $^\circ$).

Cg_1 is the centroid of the ring S1, C8, C19, C11, C13; Cg_2 is the centroid of the ring N2, N3, C7—C10.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C1—H1 $\cdots Cg_2^i$	0.98	2.88	3.62 (2)	133
C2—H2A $\cdots Cg_1^i$	0.97	2.68	3.56 (2)	151
C20—H20 $\cdots Cg_2^{ii}$	0.93	2.77	3.58 (2)	147

Symmetry codes: (i) $-x + 1, -y, -z + 2$; (ii) $-x, -y, -z + 2$.

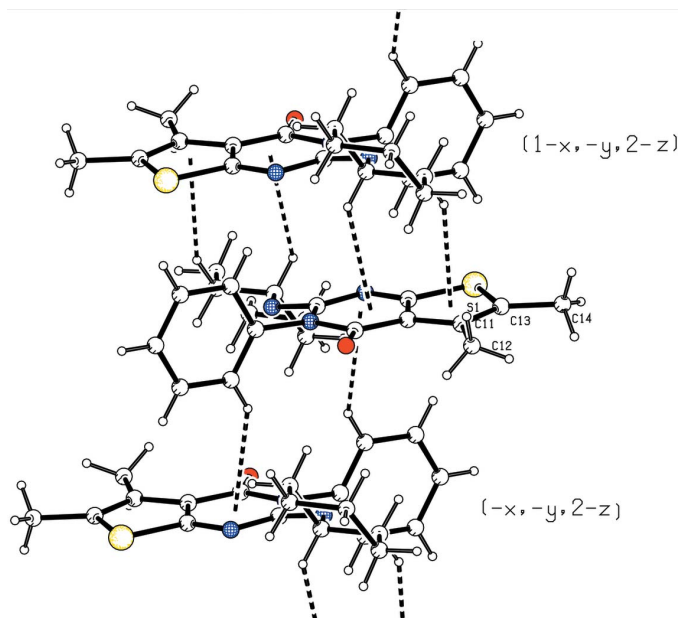


Figure 2

A view of the C—H $\cdots\pi$ hydrogen-bond stacking interactions (dashed lines).

All H atoms were located in difference maps and treated as riding atoms, with C—H = 0.93 (aromatic), 0.96 (CH₃), 0.97 (CH₂) and 0.98 \AA (CH), and N—H = 0.86 \AA , and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C}, \text{N})$ or $1.5U_{\text{eq}}(\text{methyl C})$.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON (Spek, 2003); software used to prepare material for publication: SHELXTL (Sheldrick, 2001).

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